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Catalysis is the process of increasing the rate of a chemical reaction by adding a substance known as a catalyst (*katalisator*), which is not consumed in the catalyzed reaction and can continue to act repeatedly. Because of this, only very small amounts of catalyst are required to alter the reaction rate in principle. In general, chemical reactions occur faster in the presence of a ...

Catalysis - Wikipedia

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ACS Catalysis (ACS Publications)

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Haber process - Wikipedia

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Mechanisms of catalyst deactivation - ScienceDirect

The Bend+Libration Combination Band Is an Intrinsic, Collective, and Strongly Solute-Dependent Reporter on the Hydrogen Bonding Network of Liquid Water

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